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$11 \times N = 14$ ;  $3 \times H = 3$ ;  $1 \times S = 32$ ;  $3 \times O = 16 \times 3 = 48$ . 2Empirical mass =  $14 + 3 + 32 + 48 = 97$ . 3Divide molar mass by empirical mass:  $97.01/97 = 1$ , therefore molecular formula = empirical formula. 42© Pearson Education Ltd 2015. Copying permitted for purchasing institution only. This material is not copyright free.

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